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Compelling hydrochemical evidence about limestone being dissolved as a result of the H₂S-induced acidity of the thermal brine that supplies Hercules karst spring

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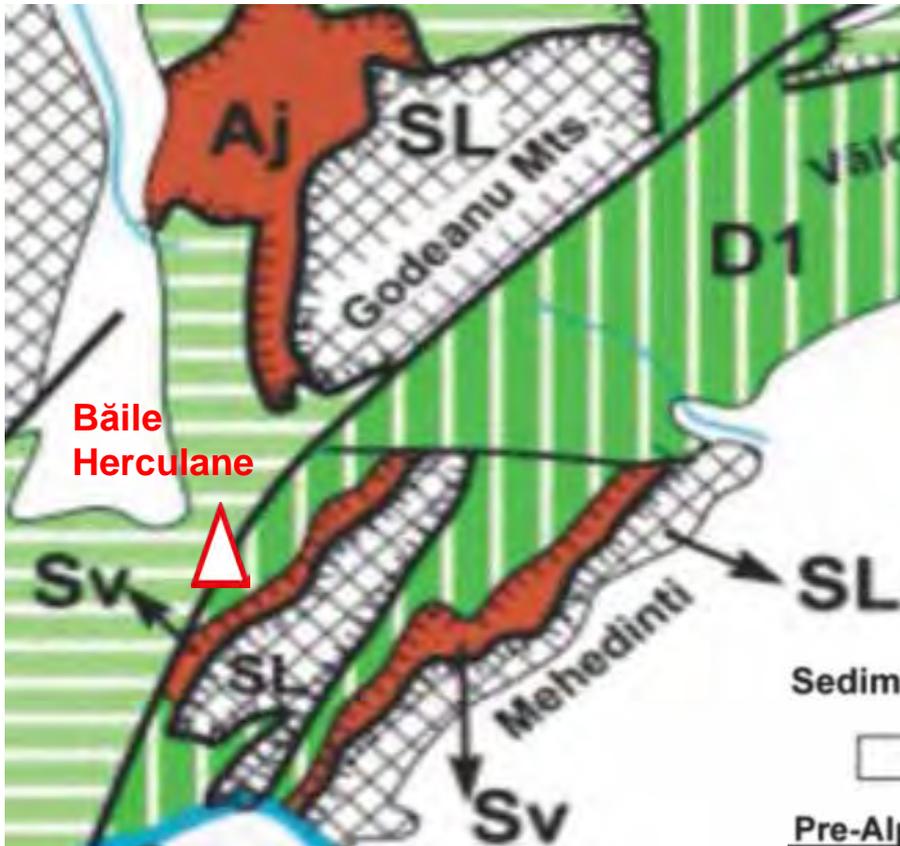
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The present study addresses

**hydrochemical monitoring of
4 thermal water outflows (31-54 °C)
at Băile Herculane, over a period of
about 3 months:
*mid-November 2013 to mid-February
2014***





A complicated nappe-structure in the Southern Carpathians

The thermal water discharges from **Upper Danubian nappes** formations

In that area, the **Cerna-Jiu major strike-slip fault** brings **Upper** and **Lower Danubian** nappes in lateral contact

Băile Herculane



Sedimentary cover

Post-Paleocene, Cenozoic cover

Pre-Alpine terrains of the Getic-Supragetic basement

SL
Sebeş-Lotru

Severin-Arjana - nappes

Aj Sv
Arjana nappe Severin nappe

Danubian

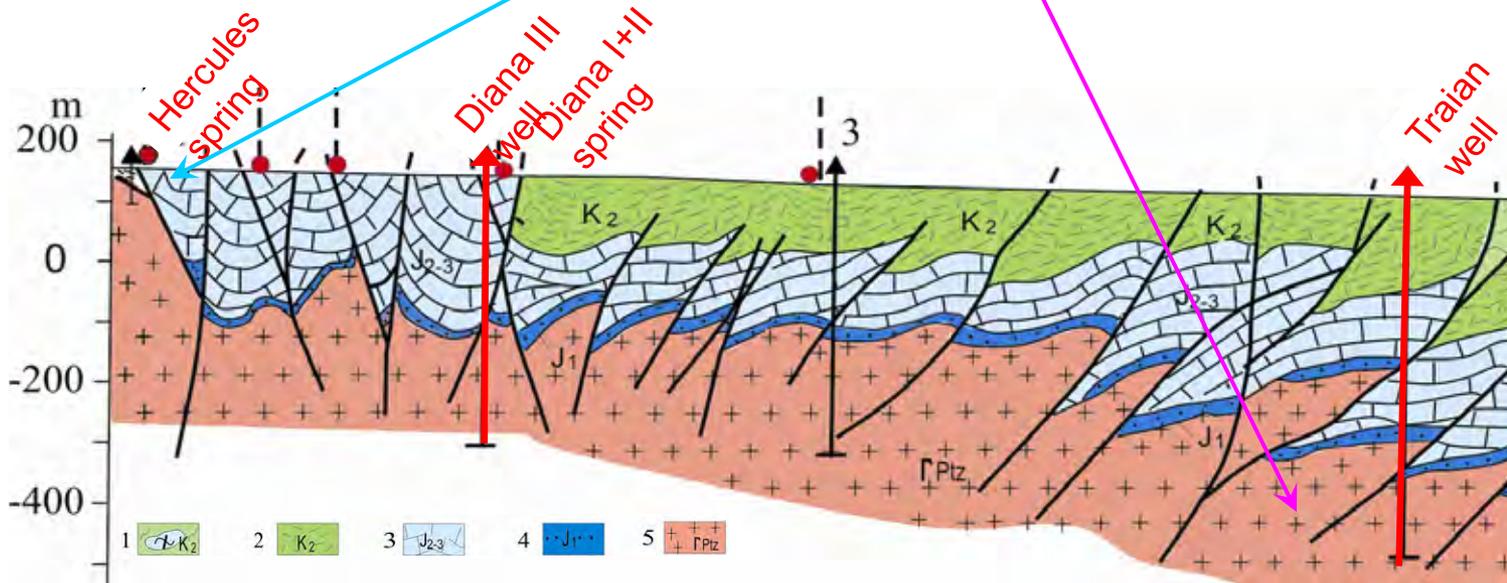
D1 D2
Lower Danubian Upper Danubian

0 5 10 15 20 Km

Stremţan, 2014

Reservoir rocks:

- Middle-Late Jurassic limestone
- Variscan-age Cerna granite



Povară et al., 2015

1. Late Cretaceous wildflysch; 2. Middle Cretaceous turbiditic limestone;
3. Middle-Late Jurassic limestone; 4. Liassic sandstone; 5. Variscan granite

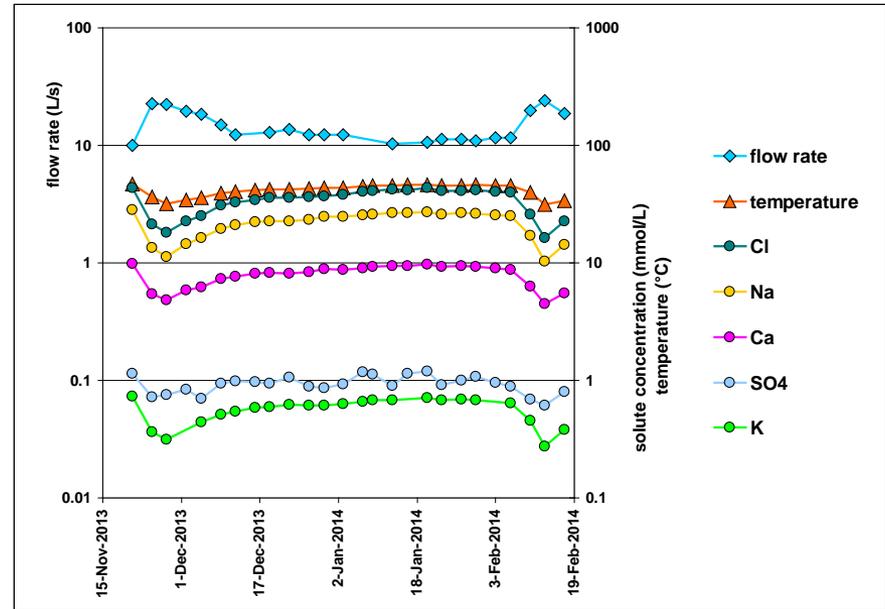
The thermal water outflows – two distinct categories:

1. of *highly variable* discharge:
Hercules karst spring
2. of *quasi-constant* and *relatively low* discharge:
Diana III well
Diana I+II spring
Traian well

Correspondingly:

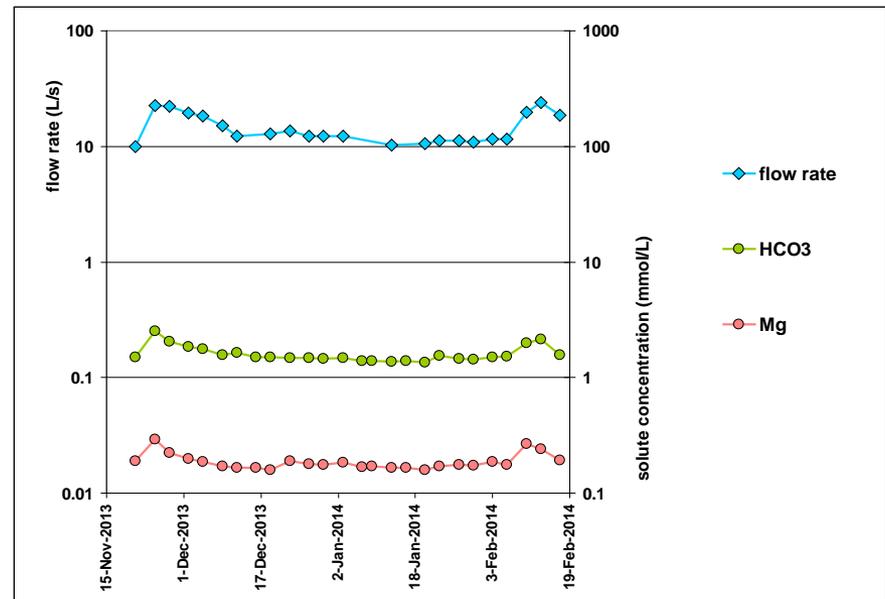
for Group 1 outflow (**Hercules spring**) both **temperature** and the **main chemical components**' concentrations **fluctuate significantly**:

temperature, Cl^- , Na^+ , Ca^{2+} , SO_4^{2-} and K^+ opposite to the flow rate



and

Mg^{2+} and HCO_3^- concordantly to the flow rate



Sampling frequency adopted for this type of behavior: twice a week

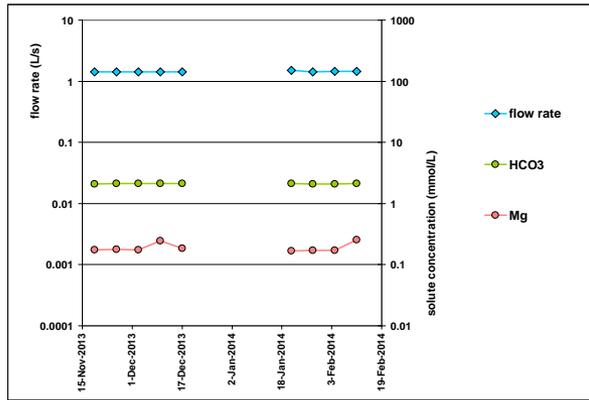
for Group 2 (**non-karst**) outflows:

quasi-insignificant fluctuations of both temperature, and of the main chemical components' concentrations

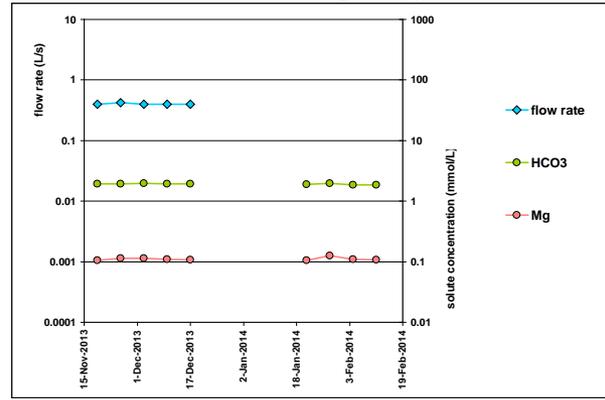
Sampling frequency adopted for this type of behavior:
once a week

(yet no sampling during periods of spa activity - which involved repeated maneuvers of well-bore valves switching: this process seemingly induced disturbances in the discharged water chemistry)

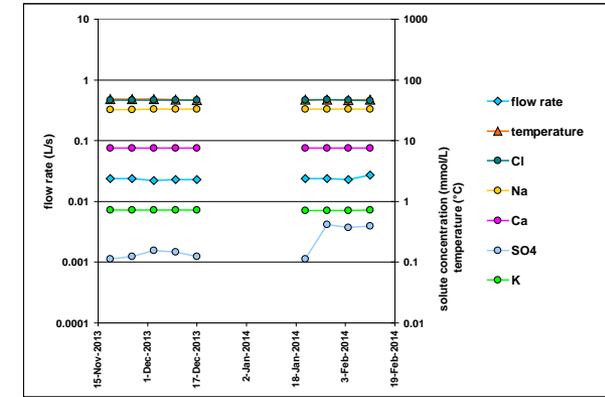
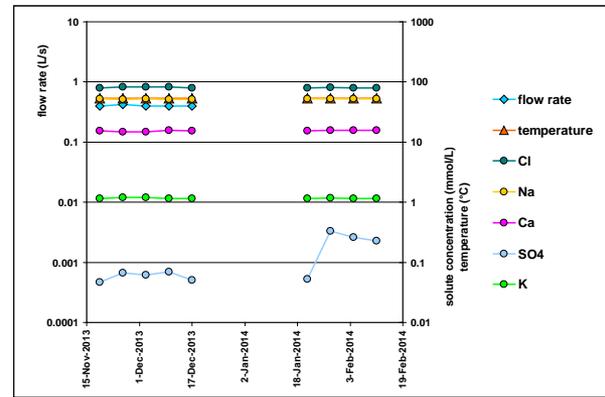
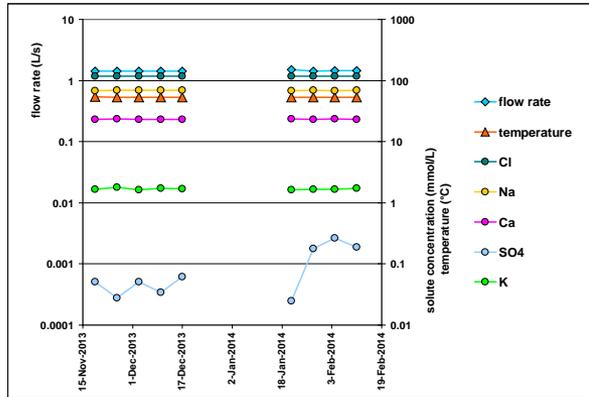
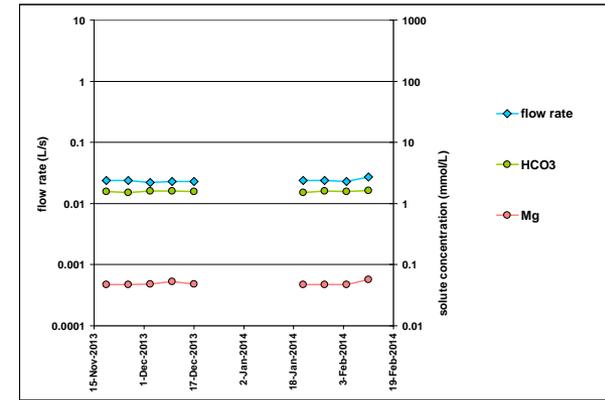
Traian well



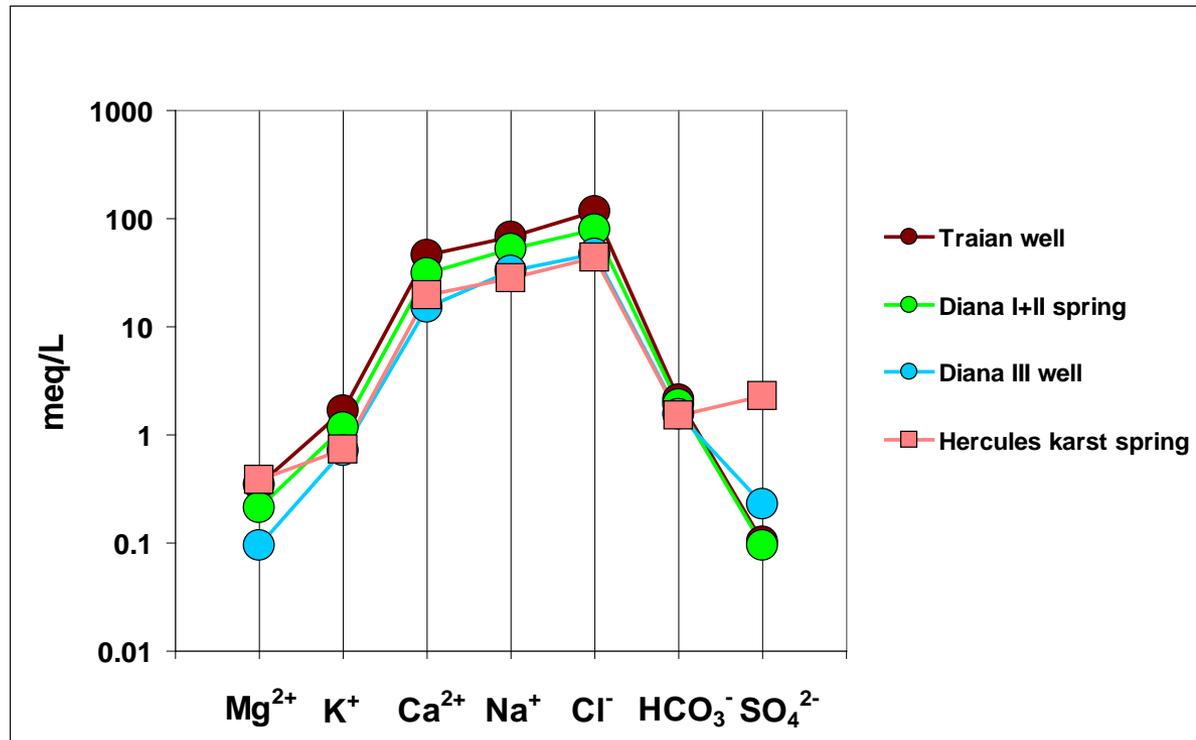
Diana I+II spring



Diana III well



**In spite of the contrasting hydrological behavior -
all those outflows have a very similar general chemical facies : Na-Ca-Cl**



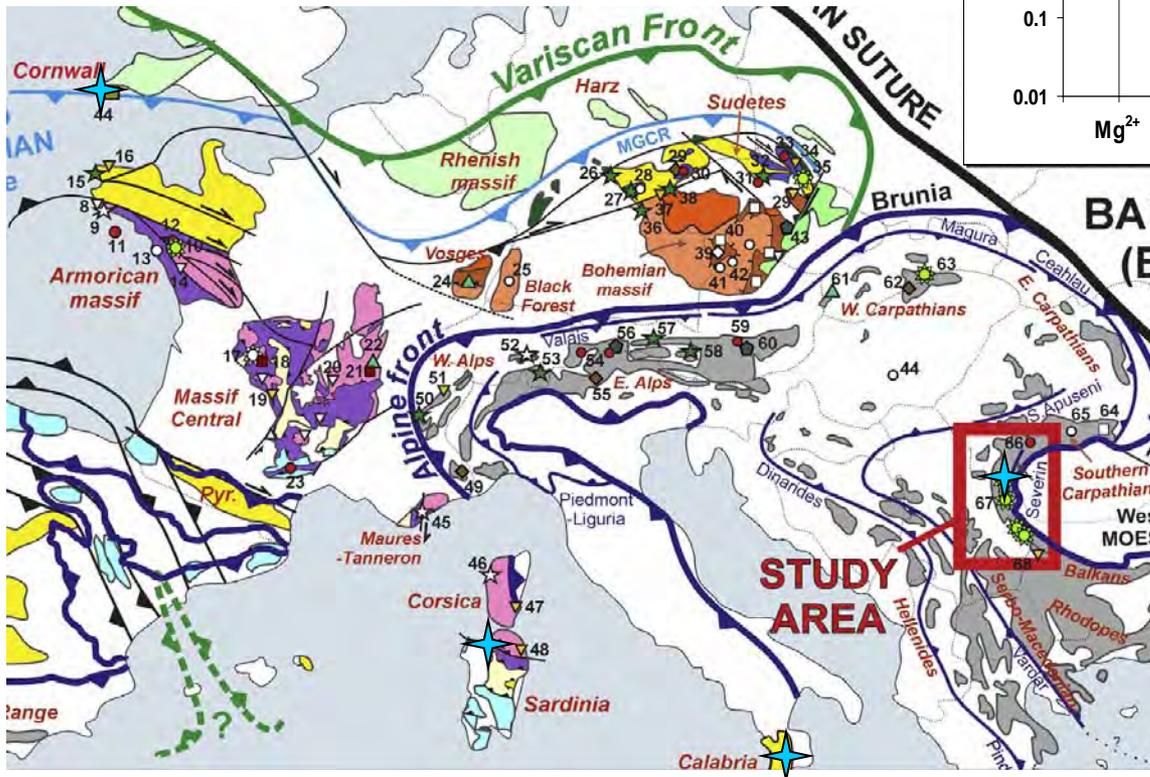
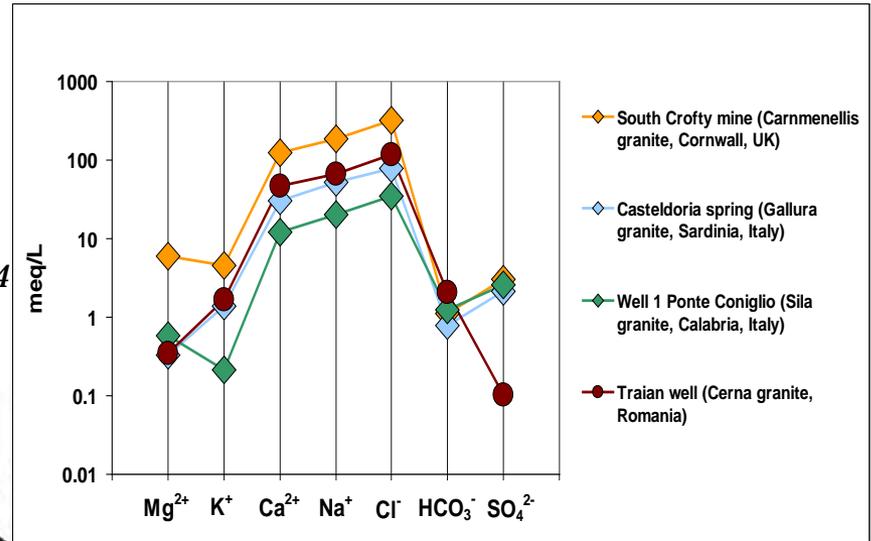
**It was therefore inferred that some kind of mixing between:
a Na-Ca-Cl water
and
weakly mineralized fluids
operated within the entire geothermal system of Băile Herculane.**

The highly mineralized Na-Ca-Cl water origin:

likely, the **Variscan-age Cerna granite**.

Very similar - Na-Ca-Cl - chemical facies - also other thermal waters discharging from Variscan-age granites in Europe:

Carmenellis granite (Cornwall, UK) – Edmunds et al., 1984
Gallura granite (Sardinia, Italy) – Angelone et al., 2005
Sila granite (Calabria, Italy) – Apollaro et al., 2016



Variscan map of Europe
(Plissart et al., 2017)

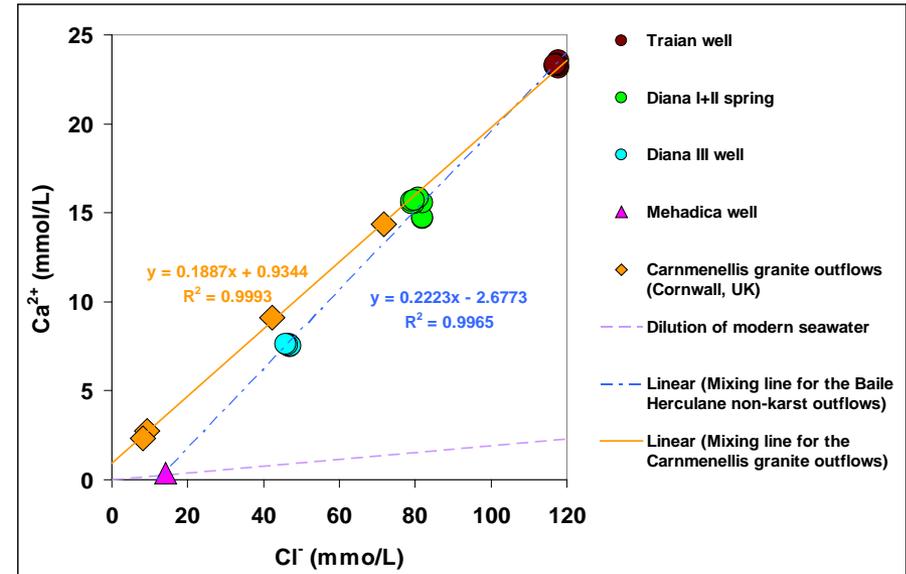
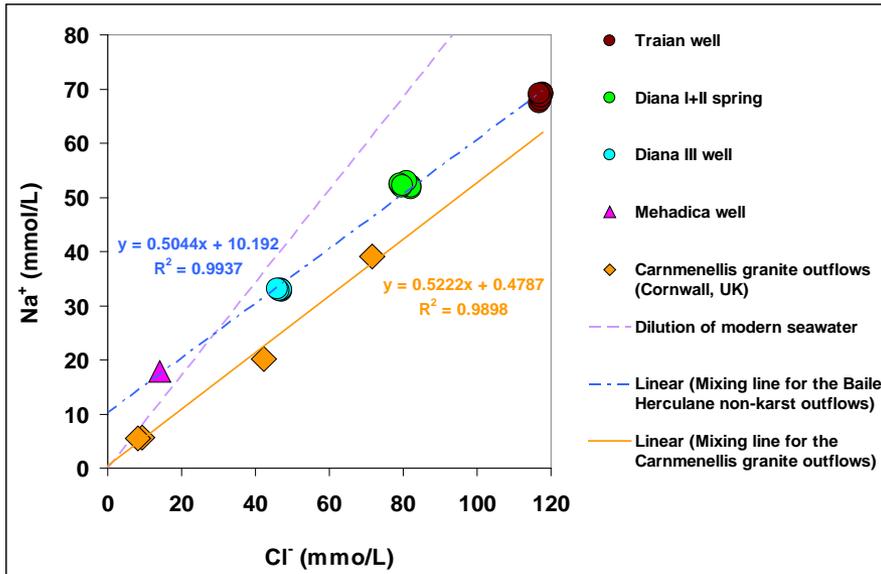
The diluting fluid(s) origin ?

There are considered **conservative natural tracers**; (“**conservative**” - negligible chemical interaction with the host rock)

The most widely used conservative natural tracer – **chloride (Cl⁻)**

In the Băile Herculane **non-karst outflows** - also **Na⁺** and **Ca²⁺** behave conservatively (similar behavior was also recorded for **Na-Ca-Cl outflows** discharging from the **Carmmenellis granite**, UK - *Edmunds et al., 1984*)

The concentration of one conservative tracer is plotted against the concentration of the other conservative tracer (“**reciprocal concentration plots**”):



Since both tracers in a diagram are conservative, **each sample** (data-point) is assumed to mirror **ONLY** a specific **mixing ratio** between two **parent waters**.

“**mixing line**” – straight line connecting the **two parent waters** (each of them of assumedly constant concentration) which **mix** in order to result the **sampled fluid**;

For the **mixing line** corresponding to the **non-karst outflows** (Traian, Diana I+II, Diana III), a **proxy** for the **diluting parent-water** could be the **Mehadica well** thermal (**23.6°C**) **outflow**.

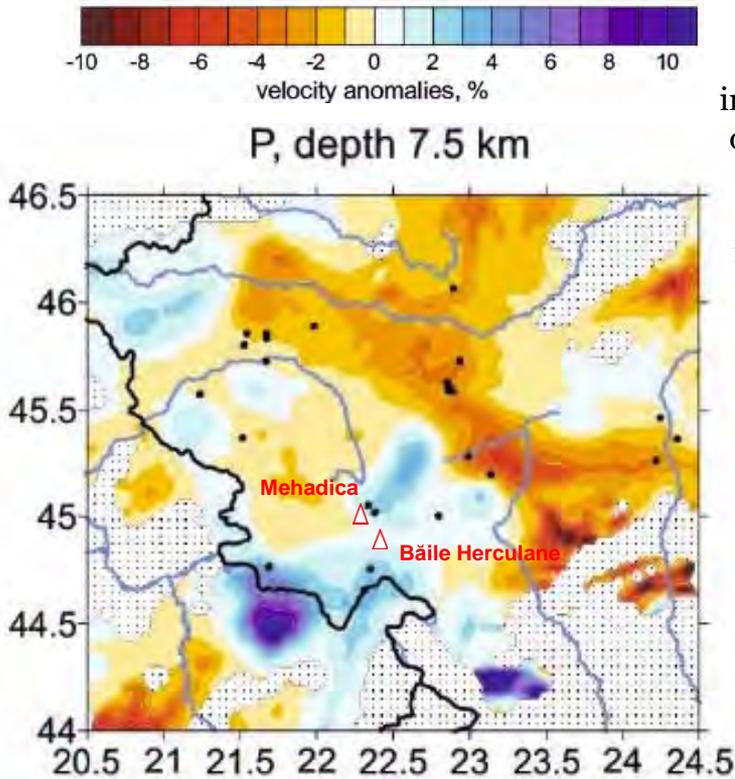
Mehadica well:

located 20 km away from Băile Herculane, in a Neogene, post-nappe sedimentary basin

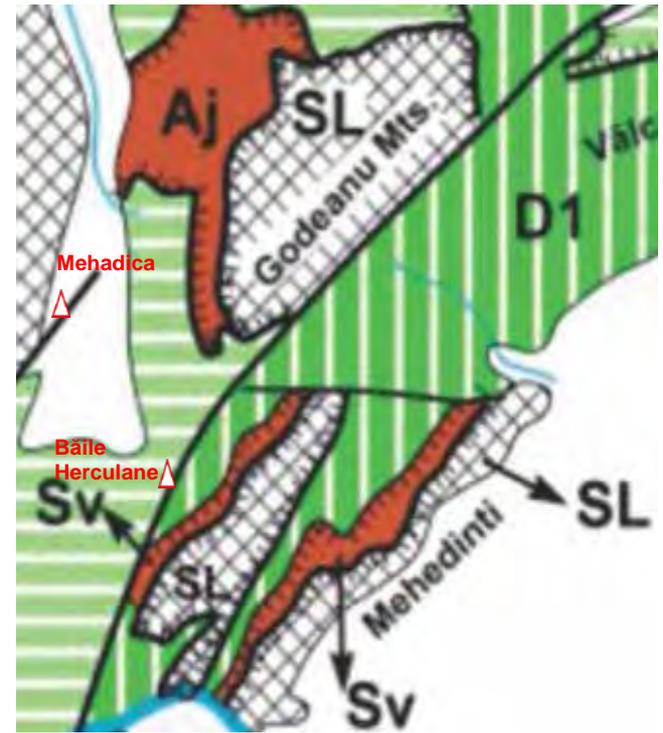
- no information about depth, completed interval, tapped aquifer(s)

A connection between “Mehadica-type” water, and the Băile Herculane outflows appears unlikely, given:

- the different basement units outcropping next to the two regions
- the general SW-NE strike of the major tectonic structures



Yet recent seismic tomography clearly indicates a WNW-ESE oriented low-velocity structure (intensely fractured basement region ?) connecting Mehadica and Băile Herculane



Sedimentary cover

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Pre-Alpine terrains of the Getic-Supragetic basement

SL
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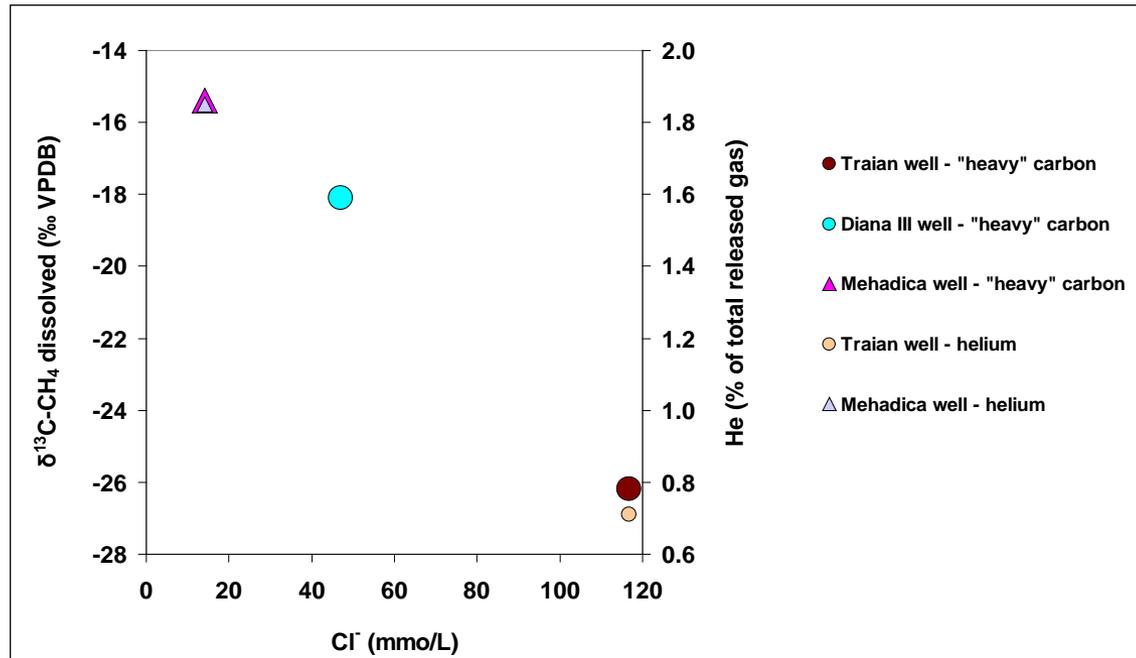
D1 D2
Lower Danubian Upper Danubian

Stremțan, 2014

Additional support for the inferred **mixing relationship** involving the **Mehadica well** and the **non-karst outflows** at Băile Herculane:

the **associated gas composition**

The **“Mehadica-type” fluid** (which progressively dilutes, at Băile Herculane, the brackish Na-Ca-Cl parent-water) has a corresponding **contribution** - with **He** and with **abiotic-origin CH₄** - to the gas phase associated to the **non-karst outflows**



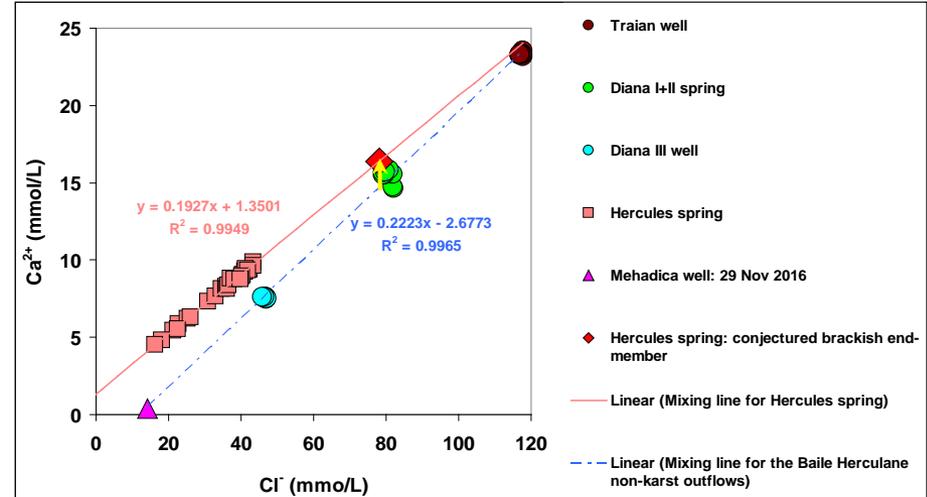
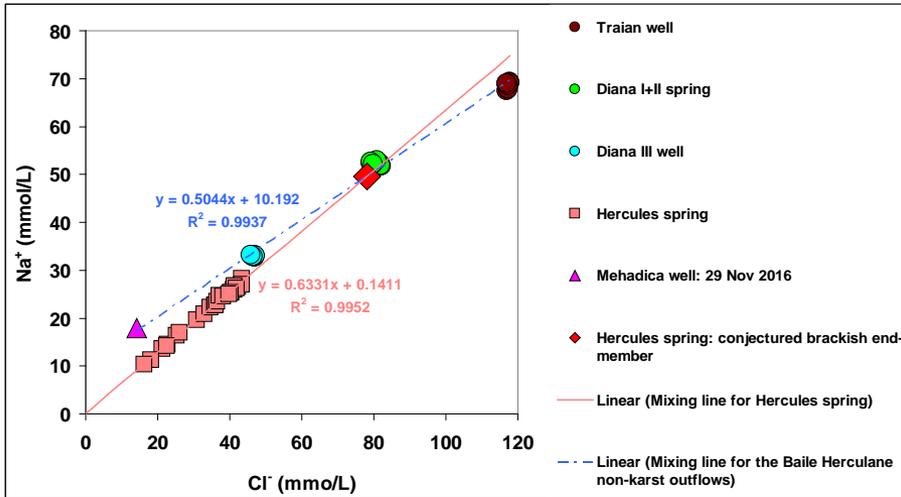
- contribution of **abiotic methane** (mirrored by **large ¹³C concentrations**) to the total content of dissolved CH₄ (*Ionescu et al., 2017*): progressively **increases from Traian well, toward Mehadica well** (while, correspondingly, the thermogenic methane contribution, decreases)
- **He** percentage in the total released gas (*Cosma & Ristoiu, 1999*) also **increases from Traian well, toward Mehadica well**

For the **non-karst outflows**:

the **mixing ratios vary** mainly **from one sampling site to another** (while each such separate sampling site displays virtually no time-fluctuations of the concerned solute contents): therefore, mixing lines are fitted to the data-points comprehensively aggregated for all the dates when the non-karst outflows are sampled

For the **Hercules karst spring**:

- the **mixing ratios vary between** samples collected **at different time-moments (different mixing ratios** between the **brackish parent-water**, and the **meteoric-derived inflow of karst freshwater**)



In the Na vs Cl reciprocal concentration plot, the two mixing lines intersect at a Cl⁻ concentration of about 78 mmol/L, suggesting that:

- the brackish end-member which supplies Hercules spring originates in a mixing process which involves the same end-members as those which contribute to the non-karst outflows;

and

- in the case of the brackish end-member which supplies Hercules spring, the mixing ratio between those two end-members is roughly the same as the one displayed by the Diana I+II spring.

But the Ca²⁺ vs. Cl⁻ mixing lines intersection occurs at a Cl⁻ concentration of about 136 mmol/L, suggesting that:

- although originating in the binary mixing process outlined for the non-karst outflows, the brackish parent-water which supplies Hercules spring should be more concentrated even than the Traian well discharge.

One possible explanation:

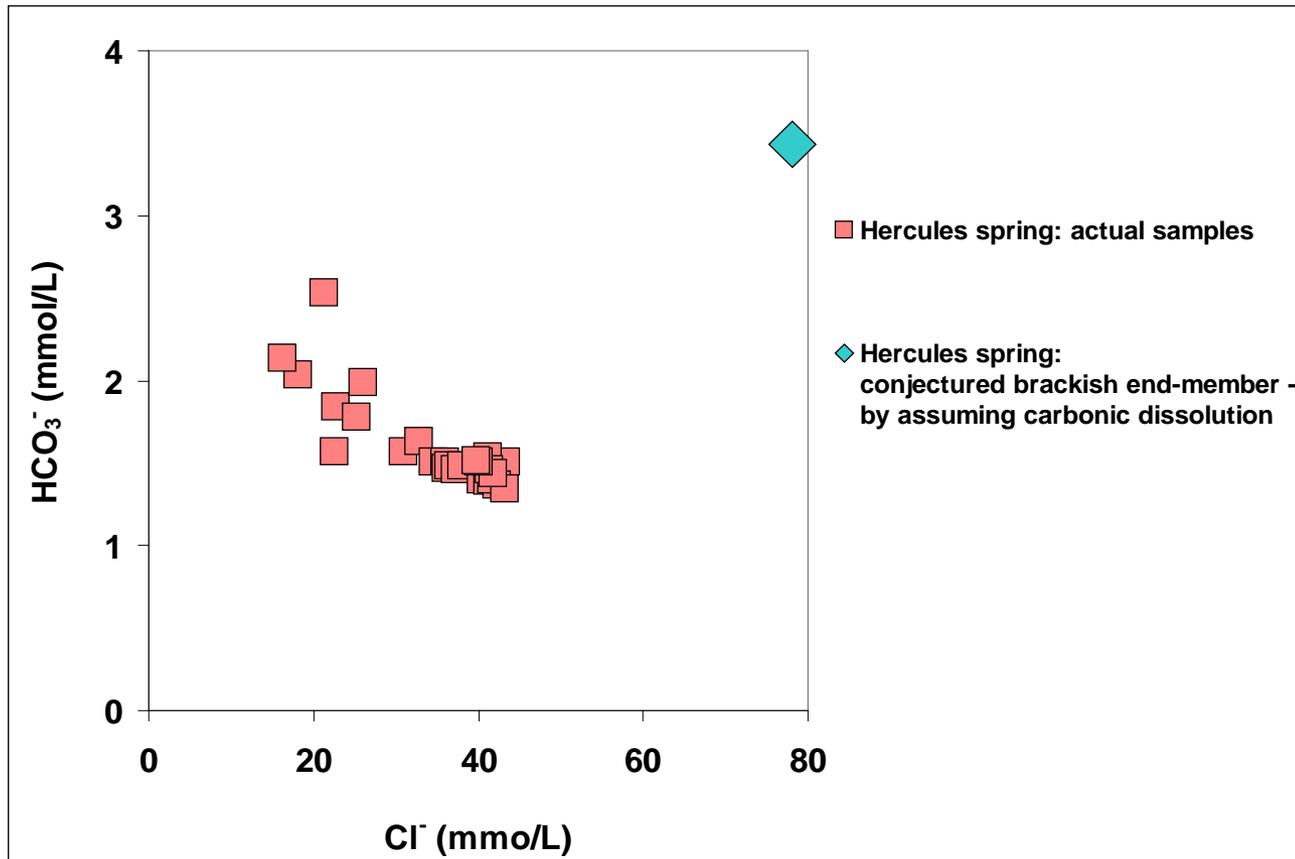
The brackish parent-water which supplies Hercules spring indeed has a Cl⁻ concentration of about 78 mmol/L; but its Ca²⁺ content is larger than the one corresponding simply to the non-karst outflows mixing trend.

The accordingly conjectured **Ca²⁺ enrichment** (about **1.7 mmol/L**) – **likely**, a result of **limestone dissolution**.

Yet **NOT** the (habitually-encountered) **limestone attack by CO₂ and H₂O**:



The dissolved HCO₃⁻ amount which would result ensuing to the corresponding carbonic dissolution (3.4 mmol/L, needed for balancing the brackish parent-water Ca²⁺ enrichment of 1.7 mmol/L) plots significantly faraway from the HCO₃⁻ vs. Cl⁻ trend outlined by the Hercules spring data-points.



Additional clue - provided by the **fate of H₂S**

For the **non-karst outflows** - H₂S concentrations in the range **0.85 - 1.73 mmol/L**

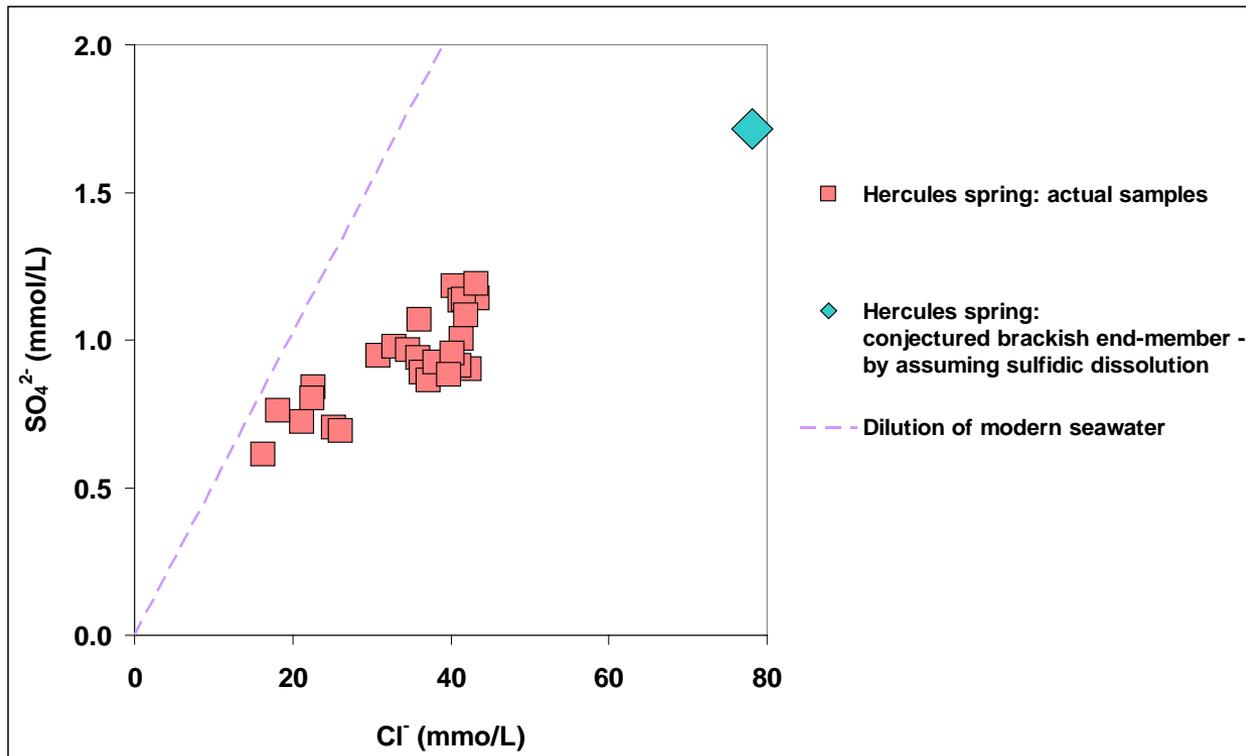
For **Hercules spring** - H₂S concentration always **below the detection limit** (b.d.l.), of **2.9 x 10⁻³ mmol/L**

H₂S concentrations (mmol/L)

Sampling date	Traian well	Diana I+II spring	Diana III well	Hercules spring
19-Nov-2013	1.616	1.348	0.948	
21-Nov-2013				b.d.l.
25-Nov-2013				b.d.l.
26-Nov-2013	1.534	1.206	1.020	
28-Nov-2013				b.d.l.
02-Dec-2013				b.d.l.
03-Dec-2013	1.509	1.301	1.015	
05-Dec-2013				b.d.l.
09-Dec-2013				b.d.l.
10-Dec-2013	1.331	1.283	0.965	
12-Dec-2013				b.d.l.
16-Dec-2013				b.d.l.
17-Dec-2013	1.560	1.049	1.006	
19-Dec-2013				b.d.l.
23-Dec-2013				b.d.l.
27-Dec-2013				b.d.l.
30-Dec-2013				b.d.l.
03-Jan-2014				b.d.l.
07-Jan-2014				b.d.l.
09-Jan-2014				b.d.l.
13-Jan-2014				b.d.l.
16-Jan-2014				b.d.l.
20-Jan-2014				b.d.l.
21-Jan-2014	1.506	1.602	1.021	
23-Jan-2014				b.d.l.
27-Jan-2014				b.d.l.
28-Jan-2014	1.397	1.469	0.927	
30-Jan-2014				b.d.l.
03-Feb-2014				b.d.l.
04-Feb-2014	1.665	1.605	0.852	
06-Feb-2014				b.d.l.
10-Feb-2014				b.d.l.
11-Feb-2014	1.730	1.378	0.985	
13-Feb-2014				b.d.l.
17-Feb-2014				b.d.l.

Still likely, the **brackish end-member** involved in the **Hercules spring** supply had, **initially, comparably large amounts of H₂S**.

The latter were able to **provide, by oxidation**, the **1.7 mmol/L of SO₄²⁻** needed **for balancing the Ca²⁺** which – assumedly - was **up-taken into solution** ensuing to the **limestone attack by the dissolved H₂S**



This **SO₄²⁻ content estimated** for the **brackish end-member (1.7 mmol/L)** is also **consistent with the SO₄²⁻ vs. Cl⁻ trend** outlined for Hercules spring.

And because of the complete oxidation to SO₄²⁻, virtually no H₂S remained in the Hercules spring discharge

What is the resulting **limestone dissolution rate**?

karst freshwater supplying Hercules spring - has negligible Cl^- input;

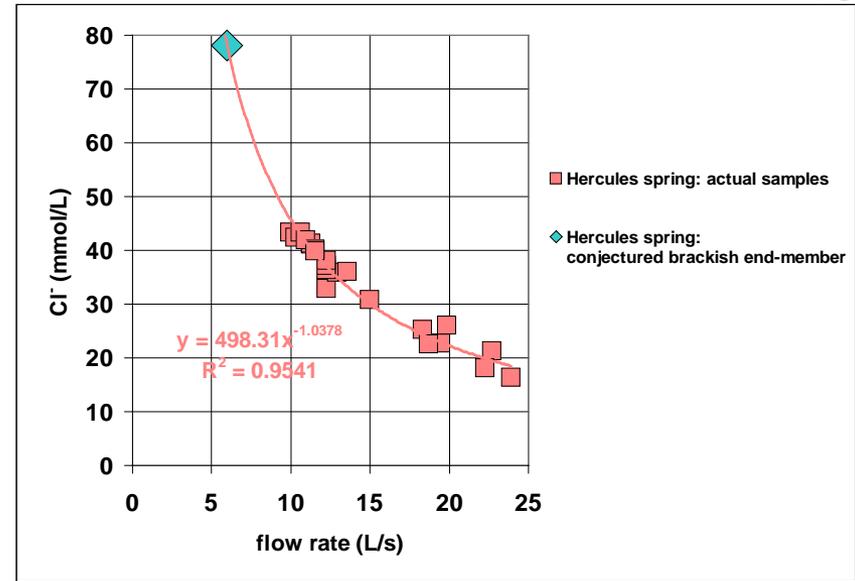
hence

Hercules spring output of Cl^- ($Q_o c_o$) is fully balanced by the Cl^- input due to the brackish parent-water ($Q_b c_b$),

namely:

$$c_o = Q_b c_b Q_o^{-1}$$

If a **power-law** with an **exponent** ≈ -1 can be **fitted** to the c_o vs Q_o plot, this indicates that the product $Q_b c_b$ is **constant**.



Q_o – total flow rate discharged by Hercules spring
 c_o – Cl^- concentration recorded at Hercules spring
 Q_b – brackish parent-water flow rate
 c_b – brackish parent-water Cl^- concentration

c_b itself is also **constant** (otherwise no tight mixing lines could have been fitted to the conservative tracers' concentrations); hence Q_b **must be constant** as well.

Assessing Q_b :

the **power-law line** is **extrapolated** to the **Cl^- content** of about **78 mmol/L** (estimated for the **brackish parent-water which supplies Hercules spring**)

It results: $Q_b = 6 \text{ L/s}$ – the **brackish parent-water flow rate**

The amount of Ca^{2+} **up-taken** by **sufidic dissolution** and **carried by this flow rate**: $1.7 \text{ mmol/L} \times 6 \text{ L/s} = 10.2 \text{ mmol/s}$

The amount of correspondingly **dissolved CaCO_3** is also **10.2 mmol/s** - namely $\approx 1 \text{ g/s}$, or $\approx 2.6 \text{ tons/month}$, or $\approx 1 \text{ m}^3/\text{month}$.

In fact, the **limestone amount dissolved by sufidic attack** is likely **double** ($\approx 2 \text{ m}^3/\text{month}$) – since a mine gallery having been excavated in the late 1980'ies, currently diverts half of the discharge that was originally flowing through Hercules karst spring.

CONCLUSIONS

It had been conjectured - since long time ago, e.g. *Popescu-Voitești, 1921* - that the **Băile Herculane thermal water composition** resulted from **mixing** between several (three?) **different water types**

And it was quite reasonable to assume that the **H₂S-bearing waters** were **dissolving limestone**

So, what's new about that?

Actually, **no quantitative evidence** had been **so far** provided **for any of those two inferences**.

And accordingly, **no quantitative estimates** had been issued **for** the involved amounts – namely:

- **solute concentrations** of **contributing parent-waters**;

and

- **sulfidic dissolution rate**

Acknowledgements

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